

Review

# Natural Sweeteners: Biosynthesis and Pharmacological

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**Abstract:** Natural sweeteners have gained attention as promising alternatives to artificial sweeteners and traditional sugars due to their low or zero-calorie content and diverse bioactive properties. This review examines several key natural sweeteners, including steviol glycosides, mogrosides, glycyrrhizin, neohesperidin, and fructooligosaccharides. The biosynthetic pathways of these sweeteners, combined with their health benefits, make them valuable in managing metabolic disorders like diabetes and obesity. This article also discusses their potential applications in the food, pharmaceutical, and nutraceutical industries, highlighting their role in promoting healthier dietary practices. Further research is needed to optimize their production and fully assess their long-term health impacts, which will be essential for unlocking their therapeutic potential.

**Keywords:** sweetener, steviol glycoside, mogroside, glycyrrhizin, neohesperidin, fructooligosaccharides

## 1. Introduction

In recent years, natural sweeteners have garnered significant attention due to their potential health benefits and wide applications in the food and pharmaceutical industries. Unlike traditional artificial sweeteners and high-calorie sugars, natural sweeteners not only provide healthier low- or zero-calorie alternatives but also exhibit a range of bioactive properties, including antioxidant, anti-inflammatory, and anti-diabetic effects, which make them particularly advantageous for managing metabolic disorders. Common examples of natural sweeteners include steviol glycosides derived from *Stevia rebaudiana*, mogrosides from *Siraitia grosvenorii* (monk fruit), glycyrrhizin from licorice, neohesperidin, and fructooligosaccharides (Table 1). These sweeteners are extensively utilized in foods and beverages for their sweetness, while also demonstrating significant potential in regulating blood glucose levels, improving metabolic health, and reducing inflammation. As awareness of metabolic diseases such as diabetes and obesity continues to rise, the role of natural sweeteners in promoting a healthier diet is becoming increasingly relevant. This review will explore the biosynthetic pathways and pharmacological activities of these natural sweeteners, highlighting their potential for future health applications.

**Table 1.** Five common natural sweeteners.

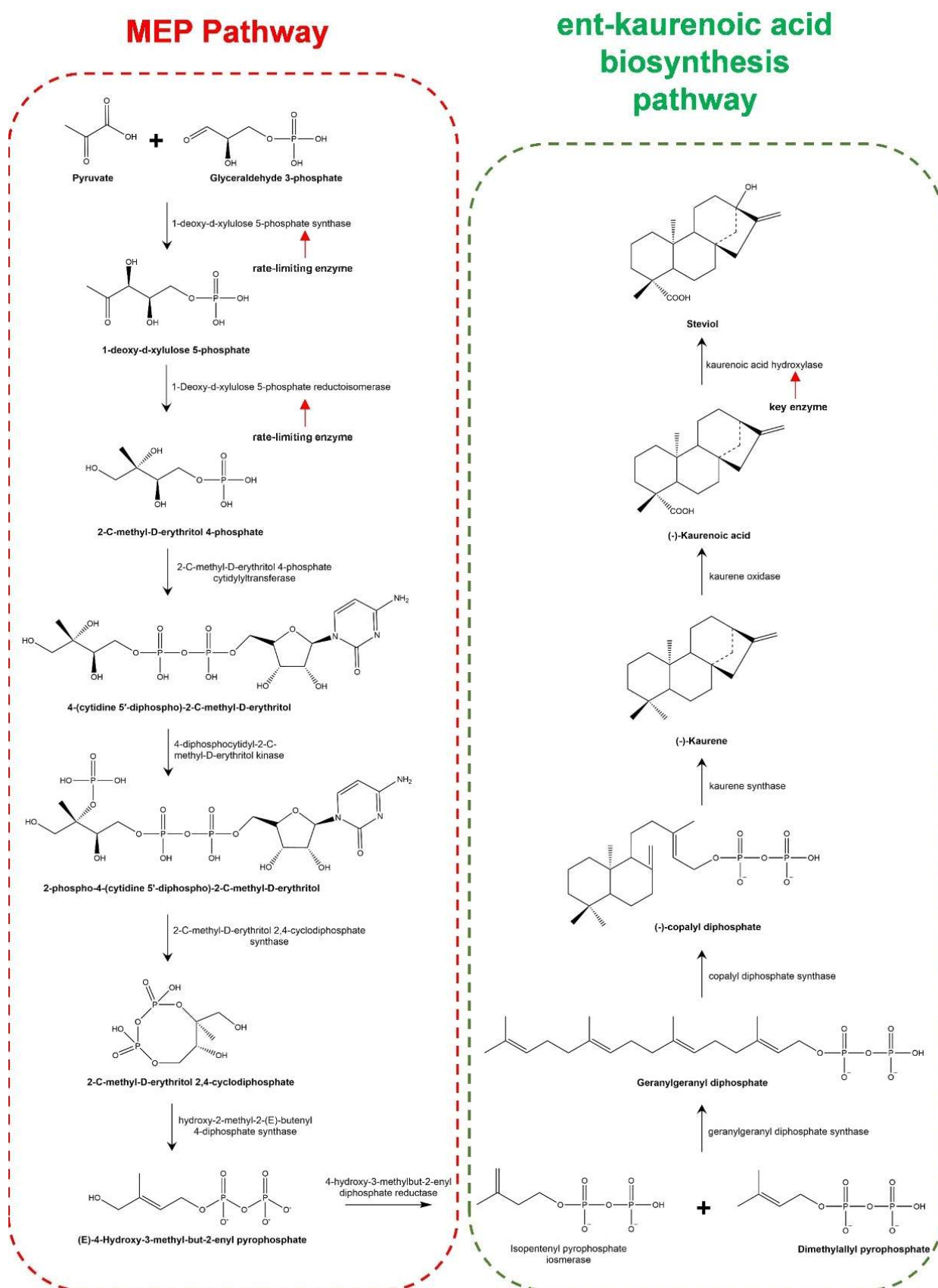
Compound Name	Category	Source	Relative Sweetness to Sucrose
Steviol Glycosides	Glycosides	Stevia leaves ( <i>Stevia rebaudiana</i> )	30–320 times
Mogrosides	Triterpene Glycosides	Monk fruit ( <i>Siraitia grosvenorii</i> )	100–250 times
Glycyrrhizin	Triterpene Glycosides	Licorice root ( <i>Glycyrrhiza glabra</i> )	50 times
Neohesperidin	Flavonoids	Citrus peels Natural fruits,	\
Fructooligosaccharides	Oligosaccharides	vegetables, and plant fibers	0.3–0.5 times

## 2. Steviol Glycosides

Steviol glycosides (SGs) are natural sweeteners derived from the leaves of the South American plant *Stevia rebaudiana*. They are widely utilized in food and beverages as low- or zero-calorie sweeteners. In addition to *Stevia rebaudiana*, related species such as *S. phlebophylla* and the Rosaceae plant *Rubus chingii* also contain these compounds. SGs are reported to be 30 to 320 times sweeter than sucrose, and they are not metabolized by the human body, meaning they do not trigger a glycemic response. This property makes them suitable for individuals with diabetes or those managing blood sugar levels. However, the taste profile of SGs varies depending on the specific glycoside. For instance, rebaudioside M is recognized for its favorable taste profile compared to stevioside and rebaudioside A, which may sometimes exhibit a bitter aftertaste. The primary SGs found in Stevia leaves include stevioside (5–10%), dulcoside A (0.5–1%), rebaudioside A (2–4%), and rebaudioside C (1–2%). Commercially available SG mixtures typically consist of approximately 80% stevioside, 8% rebaudioside A, and 0.6% rebaudioside C. The first isolation of stevioside and rebaudioside A was accomplished by French chemists Bridel and Lavielle in 1931 [1]. The Stevia market size is estimated at USD 0.84 billion in 2024, and is expected to reach USD 1.36 billion by 2029, growing at a compound annual growth rate (CAGR) of 10.12% during the forecast period (2024–2029) [2]. Steviol glycosides are considered safe sweeteners. In 2006, following an analysis of several studies on stevia and steviol glycosides in humans and animals, the World Health Organization (WHO) stated that ‘stevioside and rebaudioside A are not genotoxic in vitro or in vivo and that the genotoxicity of steviol and some of its oxidative derivatives in vitro is not expressed in vivo [3].

## 2.1 Biosynthesis

SGs, the primary sweetening compounds in *Stevia rebaudiana*, are diterpenoid glycosides, and their biosynthesis has been extensively studied [4,5]. These glycosides are synthesized via the methylerythritol phosphate (MEP) pathway, which operates independently of the mevalonate (MVA) pathway. This discovery was made using glucose labelling to follow the biosynthetic pathway. The MEP pathway plays a crucial role in forming the diterpenoid glycoside backbone of SGs in *S. rebaudiana* [6,7]. The condensation reaction catalysed by 1-deoxy-D-xylulose 5-phosphate synthase (DXS) is the first step in the MEP pathway [8]. In this reaction, pyruvate and glyceraldehyde 3-phosphate (G3P) are converted to 1-deoxy-D-xylulose 5-phosphate (DXP) under the catalysis of DXS [9]. DXS is the rate-limiting enzyme in the MEP pathway and controls the flux through the pathway [10,11]. DXP is then reduced to MEP by 1-deoxy-D-xylulose 5-phosphate reductoisomerase (DXR). The subsequent steps in the MEP pathway involve the enzymes 4-pyrophosphate 2-C-methyl-D-erythritol synthase (CMS), 4-pyrophosphate 2-C-methyl-D-erythritol kinase (CMK) and 2-C-methyl-D-erythritol 2,4-cyclopyrophosphate synthase (MCS), which, together with (E)-4-hydroxy-3-methylbut-2-enyl pyrophosphate synthase (HDS), catalyse the complex bio-organometallic mechanism that converts MEP to (E)-4-hydroxy-3-methylbut-2-enyl pyrophosphate (HMBPP) [12]. Finally, HMBPP is reduced to isopentenyl pyrophosphate (IPP) and dimethylallyl pyrophosphate (DMAPP) by (E)-4-hydroxy-3-methylbut-2-enyl pyrophosphate reductase (HDR) through a similar mechanism [13]. The interconversion between IPP and DMAPP is facilitated by isopentenyl pyrophosphate isomerase (IDI) [11]. Following the discovery of the diterpenoid properties of steviol, the biosynthesis of the diterpene moiety of stevioside begins with the addition of IPP units to dimethylallyl diphosphate to form geranylgeranyl diphosphate (GGPP) [14]. GGPP is converted to ent-kaurene by copalyl diphosphate synthase (CPPS) and kaurene synthase (KS) through the release of energy, and then produces (-)-kaurene acid ((-)-KA) through the three-step oxidation reaction of kaurene oxidase (KO) [15]. Next, the hydroxylation reaction of ent-KA at C13 under the catalysis of kaurenoic acid 13-hydroxylase (KAH) is the key step to produce SGs [16].



**Figure 1.** The biosynthesis process of steviol.

## 2.2 Bioactive

SGs, natural non-caloric sweeteners extracted from *Stevia rebaudiana*, have demonstrated a wide range of physiological activities and are widely used in the food and pharmaceutical industries. Extensive research highlights their potential to promote metabolic health, reduce inflammation, and protect the liver. For example, SGs have been shown to alleviate insulin resistance and obesity-related metabolic disorders by improving glucose tolerance, reducing oxidative stress, and downregulating inflammatory gene expression [17]. In a separate study using high-fat diet-induced and diabetic rat models, SGs effectively regulated lipid metabolism, ameliorated hyperlipidemia, and attenuated associated organ damage [18]. Furthermore, SGs have been observed to improve glucose and lipid metabolism by modulating genes related to these pathways, showing beneficial effects in diabetic rats [19]. Similarly, in adipocyte models, SGs inhibited adipogenesis and lipid accumulation while significantly enhancing insulin sensitivity [20].

The potential metabolic benefits of SGs in humans have also been supported by several studies. A meta-analysis reported a significant reduction in fasting blood glucose levels, particularly in non-diabetic and overweight individuals, although the effect on glycated hemoglobin (HbA1c) was not statistically significant [21]. In addition, the daily consumption of beverages containing SGs did not significantly alter gut microbiota composition, but showed greater suppression of weight gain compared to sucrose consumption [22]. In terms of weight management, one study found that replacing sugar with SGs led to a significant reduction in body weight and waist circumference in overweight and pre-diabetic adults [23].

Derivatives of SGs have also shown remarkable biological activities. For example, Reb M8, synthesized by selective enzymatic glycosylation, not only exhibits enhanced sweetness, but also significantly inhibits tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ), suggesting potent anti-inflammatory effects [24]. Furthermore, in combination studies, SGs co-administered with L-arginine and trivalent chromium improved insulin sensitivity and reduced blood glucose levels in diabetic rats [25].

SGs have also been shown to have protective effects on the liver. In one study, SGs and their extracts significantly reduced radiation-induced liver damage by reducing inflammatory markers and oxidative stress [26]. Another study showed that the inclusion of SGs and dried stevia leaves improved antioxidant capacity and lipid profiles in rats [27]. In addition, SGs have been found to stimulate the release of GLP-1 via activation of bitter taste receptor signaling pathways, further enhancing their regulatory effects on glucose metabolism [28].

In summary, SGs have broad physiological activities, including benefits in diabetes management, weight control, metabolic regulation, and liver protection, making them promising candidates for various therapeutic and dietary applications.

## 3. Mogrosides

Mogrosides are triterpenoid glycosides of cucurbitane derivatives, a type of sugar substitute, found in plants such as monk fruit [29]. The total content of mogrosides in monk fruit is 3.8%, with mogroside V being the highest (0.8% to 1.3% by weight) [30]. Some mogrosides are used in traditional Chinese medicine, and some are extracted

and used as sweeteners [29]. Mogroside V extracted from monk fruit is 250 times sweeter than sucrose [29] and is expected to grow at a CAGR of 4.8% from 2023 to 2031 [2]. Unlike sugar, mogroside contributes no calories, making it an attractive alternative for people managing their weight or blood sugar levels. The process of extracting and refining mogroside is relatively complex, and there are costs associated with growing and harvesting the monk fruit (the source of mogroside). Therefore, although it is a zero-calorie, natural sweetener, it is more expensive than synthetic sweeteners such as aspartame or stevia. Mogroside's sweetness is clean and doesn't typically exhibit the bitter aftertaste associated with some other natural sweeteners such as stevia. However, depending on the formulation, it may have a slight aftertaste that some people find less desirable. In addition, Mogroside is heat stable, making it suitable for use in cooking and baking. Toxicity studies have shown that mogrosides are non-toxic and have no mutagenic, genotoxic or other adverse effects [31,32].

### 3.1 Biosynthesis

The basic structure of mogrosides, the sweet components of *Siraitia grosvenorii* (monk fruit), is a cucurbitane-type triterpenoid saponin. In recent years, the biosynthetic pathway of mogrosides, starting from acetyl-CoA, has been fully elucidated [29,33,34]. This process can be divided into the following key steps: (1) the synthesis of IPP and DMAPP via the MVA pathway; (2) the synthesis of 2,3:22,23-diepoxyqualene; (3) the synthesis of mogrol; (4) the synthesis of mogrosides.

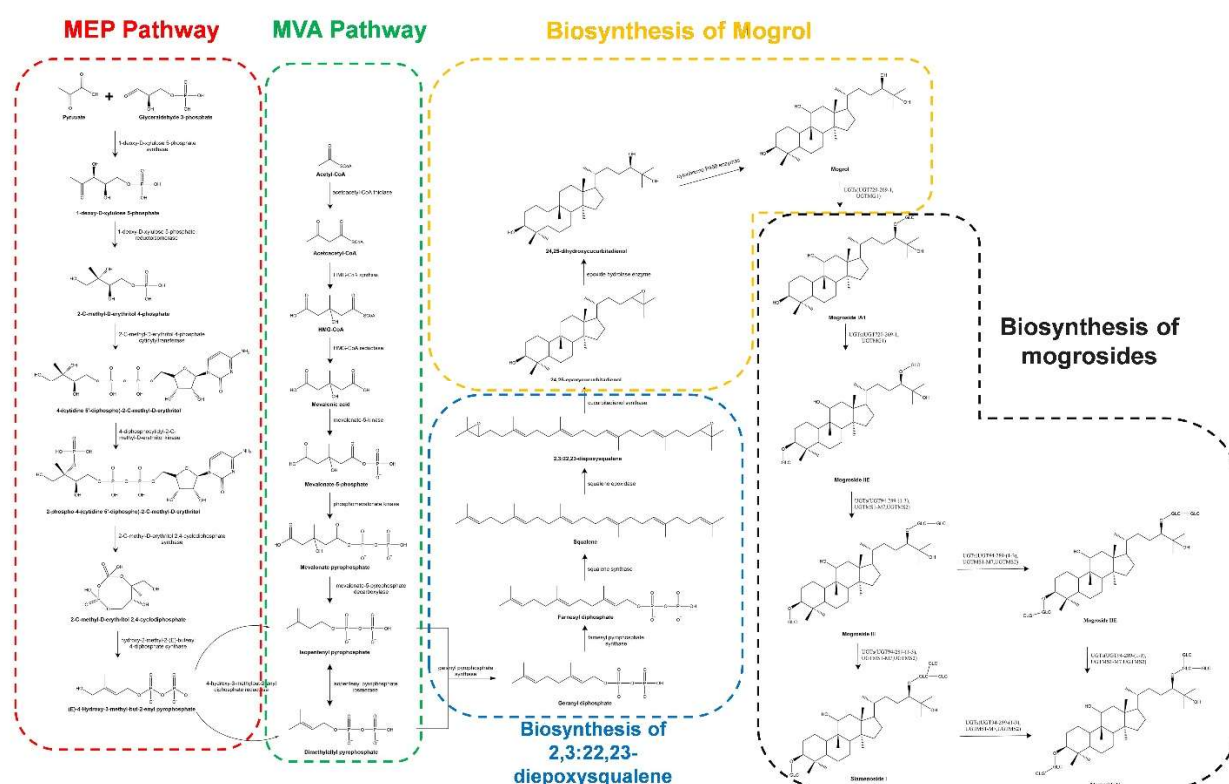
IPP and DMAPP are C5 precursors in terpene biosynthesis and can be synthesised by either the MVA or MEP pathways. In the MVA pathway, two molecules of acetyl-CoA are first condensed to form acetoacetyl-CoA, which is then combined with another molecule of acetyl-CoA under the catalysis of 3-hydroxy-3-methylglutaryl-CoA (HMG-CoA) synthase (HMGS) to form HMG-CoA. HMG-CoA is reduced to MVA by HMG-CoA reductase (HMGR), and through the sequential catalysis of mevalonate 5-kinase, phosphomevalonate kinase, and mevalonate 5-diphosphate decarboxylase, MVA is converted to IPP. IPP and DMAPP can be interconverted via isomerase activity [35]. In the MEP pathway, G3P is condensed to form DXP. DXP is then reduced to MEP by DXR. MEP is then catalyzed by 2-C-methyl-D-erythritol 4-phosphate cytidyltransferase (MCT) to produce 4-(cytidine 5'-diphospho)-2-C-methyl-D-erythritol (CD-ME). CD-ME is converted to HMBPP through a series of enzymatic actions, including phosphorylation, cyclization and ring opening. Finally, HMBPP is reduced by 4-hydroxy-3-methylbut-2-enyl diphosphate reductase to form IPP and DMAPP [36].

IPP and DMAPP undergo a head-to-tail condensation catalyzed by geranyl pyrophosphate synthase (GPS) to form geranyl diphosphate (GPP) with a C10 backbone. GPP in turn, condenses with another IPP molecule under the catalysed by farnesyl pyrophosphate synthase (FPS) to form farnesyl diphosphate (FPP), a C15 compound [37]. Squalene synthase (SQS) catalyzes the head-to-tail condensation of two FPP molecules to form squalene, which undergoes two successive epoxidation reactions catalyzed by squalene epoxidase (SE) to form 2,3:22,23- diepoxyqualene.

The synthesis of mogrol begins with the conversion of 2,3:22,23-diepoxyqualene to 24,25-epoxycucurbitadienol, catalyzed by cucurbitadienol

synthase. This intermediate is then converted to cucurbitadienol with hydroxyl groups at the C24 and C25 positions via epoxide hydrolase activity. Finally, mogrol is produced by the hydroxylation of cucurbitadienol at the C11 position by cytochrome P450 (CYP) enzymes [29].

Various UDP-glucosyltransferases (UGTs) add glucose moieties to the C3 and C24 positions of mogrol, and further glucose units are linked via  $\beta$ -1,6 and  $\beta$ -1,2 glycosidic linkages to form various branched structures, resulting in the synthesis of different mogrosides [38].



**Figure 2.** The biosynthesis process of mogrosides. GLC: glucose; UGTs: UDP-glucosyltransferases.

### 3.2 Bioactive

Mogrosides, particularly Mogroside V, have demonstrated significant therapeutic potential in various biological systems. In studies using obese **Type 2 diabetes** rat models, mogrosides were shown to significantly reduce fasting blood glucose and insulin resistance, while improving lipid profiles, thereby highlighting their role in modulating insulin secretion and activating **5' adenosine monophosphate-activated protein kinase** [39]. In addition, mogrosides positively influenced the gut microbiota in T2DM rats, leading to increased production of short-chain fatty acids, such as butyrate, which are associated with improved gut health and anti-diabetic effects [40].

In addition to its anti-diabetic properties, Mogroside V has shown potent anti-inflammatory activity. In colitis models, Mogroside V inhibited key inflammatory pathways, including MAPK-NF- $\kappa$ B and AMPK-PI3K/Akt/mTOR, suggesting its potential as a therapeutic candidate for the treatment of ulcerative colitis [40]. Similarly, in models of pulmonary inflammation, Mogroside V alleviated asthma-like

symptoms by modulating JAK-STAT and NF- $\kappa$ B signaling pathways, further highlighting its anti-inflammatory potential in respiratory diseases [41]. Furthermore, Mogroside IIE, a metabolite of Mogroside V, was shown to reduce acute lung injury by inhibiting Pla2g2a-EGFR interactions and downregulating the AKT-mTOR signaling pathway, offering potential therapeutic benefits in lung inflammation [42].

Mogroside V also demonstrated potent antioxidant properties. Studies in skin fibroblasts shown that Mogroside V reduced oxidative stress by decreasing reactive oxygen species (ROS) levels and enhancing the activity of antioxidant enzymes such as superoxide dismutase (SOD) and catalase, positioning it as a promising anti-aging agent for skincare applications [43]. Furthermore, its protective effects against hydrogen peroxide-induced oxidative stress in skin fibroblasts support its potential use in anti-aging cosmetic products [43].

In addition to these findings, Mogroside V has demonstrated potential in the treatment of complex diseases beyond diabetes and inflammation. Preclinical in silico studies have identified Mogroside V as a promising therapeutic agent for the treatment of ovarian cancer and COVID-19 by targeting critical molecular mechanisms involved in both diseases [44].

## 4. Glycyrrhizin

Glycyrrhizin is the main sweetening constituent of the root of *Glycyrrhiza glabra* (licorice). Structurally, it is a saponin. Glycyrrhizin is an extract obtained by macerating the licorice root and boiling it in water [45]. Licorice extract (glycyrrhizin) is sold in the United States as a liquid, paste, or spray-dried powder [45]. It is approved for use in certain amounts as a flavouring and fragrance in processed foods, beverages, confectionery, dietary supplements and spices [45]. It is 30 to 50 times sweeter than sucrose (table sugar) [46]. Despite the beneficial effects of liquorice, toxicity has also been reported with frequent and excessive consumption [47]. Based on in vivo studies in rats and mice, consumption of 15–229 mg/kg/day was suggested to be safe [48]. Continuous high-level exposure to glycyrrhizin is capable of producing hypermineralocorticoid-like effects; however, these effects are reversible upon withdrawal of liquorice and/or glycyrrhizin [48]. Both liquorice and glycyrrhizin have been approved for use in food products by most national and supranational regulatory agencies; however, in addition to the previously recommended dosage, the susceptibility to glycyrrhizin toxicity must be considered as it is directly related to the general health status of the individual [48,49]. The global market for glycyrrhizic acid is expected to grow at a CAGR of 6.8% between 2023 and 2030 [2].

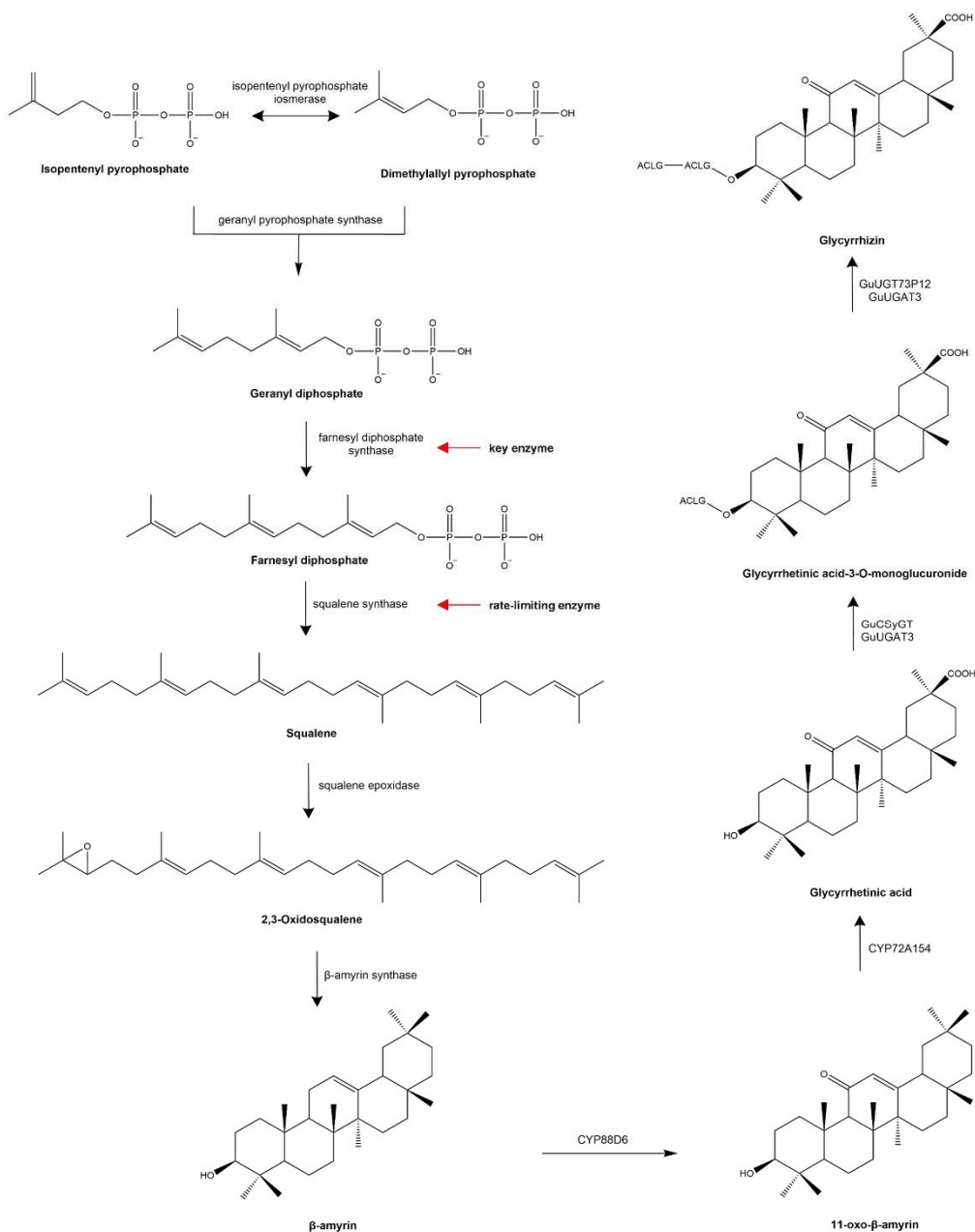
### 4.1 Biosynthesis

Glycyrrhizin is synthesised by a complex biosynthetic pathway that involves several key enzymatic steps [50]. This pathway begins with the production of isopentenyl diphosphate (IPP), an essential precursor in the biosynthesis of terpenoids.

IPP is produced by a series of enzymatic reactions [51]. Isopentenyl diphosphate isomerase (IPPI) catalyses the isomerisation of IPP to DMAPP. Under the catalysis of farnesyl diphosphate synthase (FPPS), two molecules of IPP and one molecule of DMAPP are sequentially condensed to form FPP [52]. Two molecules of FPP are then



condensed by SQS to form squalene [53]. Squalene is further converted to 2,3-oxidosqualene under the catalysis of squalene synthase [54]. 2,3-oxidosqualene is a crucial branch point in the biosynthesis of glycyrrhizin. Under the action of  $\beta$ -amyrin synthase ( $\beta$ -AS), 2,3-oxidosqualene is converted to  $\beta$ -amyrin [55]. The cytochrome P450 monooxygenases CYP88D6 and CYP72A154 catalyse two consecutive oxidation reactions at the C11 position and three oxidation reactions at the C30 position of  $\beta$ -amyrin, leading to the formation of glycyrrhetic acid [56,57]. Finally, *Glycyrrhiza uralensis* UDP-glucosyltransferase 3 sequentially transfers two glucuronide moieties to the C3 position of glycyrrhetic acid, resulting in the production of glycyrrhizin [58].



**Figure 3.** The biosynthesis process of glycyrrhizin. CYP88D6: beta-amyrin 11-oxidase; CYP72A154: 11-oxo-beta-amyrin 30-oxidase; GuUGT3: *G. uralensis* UDP-glycosyltransferase 3; GuCSyGT: *G. uralensis* cellulose synthase superfamily-derived glycosyltransferase; GuUGT73P12: *G. uralensis* UDP-glycosyltransferase 73P12; GLC: glucose.

## 4.2 Bioactive

Glycyrrhizin and glycyrrhetic acid, two major bioactive compounds derived from licorice, have demonstrated extensive therapeutic potential in various biomedical fields. Glycyrrhizin is particularly noted for its potent antioxidant properties, which result from its ability to inhibit the formation of reactive oxygen species (ROS) and scavenge free radicals, making it a valuable agent in antioxidant therapies and drug delivery systems [59]. In addition, glycyrrhizin has been shown to possess neuroprotective effects, significantly improving cognitive performance and reducing oxidative damage in animal models of cognitive impairment [60].

Glycyrrhetic acid, in contrast, has been extensively studied for its potential to enhance bioavailability and anti-inflammatory effects when formulated with solubilisers such as Soluplus®. This formulation approach has been shown to significantly enhance the therapeutic efficacy of glycyrrhetic acid without significant toxicity [61]. In addition, glycyrrhizin has the ability to form inclusion complexes and micelles, which increase the solubility and bioavailability of poorly soluble drugs, thereby improving their therapeutic efficacy [62].

Both glycyrrhizin and glycyrrhetic acid have shown potent antiviral properties, especially glycyrrhizin derivatives, which have been found to effectively inhibit the replication of viruses such as SARS-CoV and COVID-19 through interference with viral replication mechanisms and modulation of immune responses [63]. In addition, glycyrrhetic acid derivatives have demonstrated significant anti-tumor activity, particularly through their function as peroxisome proliferator-activated receptor gamma agonists, which promote apoptosis in cancer cell lines [64].

Glycyrrhizin's ability to form micelles and inclusion complexes not only enhances drug solubility but also potentiates the antioxidant activity of compounds such as carotenoids, making it a promising tool in drug delivery systems aimed at improving the stability and bioavailability of therapeutic agents [65]. This property is also instrumental in enhancing the pharmacological effects of other drugs, including anti-inflammatory and anticancer agents [66]. In addition, chemical modifications of glycyrrhetic acid have resulted in improved anti-inflammatory efficacy, offering potential benefits in the treatment of inflammatory diseases [67].

In liver-related applications, glycyrrhizin has been shown to offer significant hepatoprotective effects, including reducing liver damage, inhibiting hepatic fibrosis, and enhancing antiviral and anti-inflammatory activities [68]. Furthermore, glycyrrhizin and its derivatives exhibit broad antioxidant activity, protecting against oxidative stress by scavenging free radicals and preventing lipid peroxidation [63]. Its ability to enhance the therapeutic potential of other drugs through complex formation further underscores its utility in multiple therapeutic applications [69].

More importantly, glycyrrhizin has also shown anti-inflammatory activity *in vivo*. In a meta-analysis evaluating the anti-inflammatory activity and safety of compound glycyrrhizin in the treatment of ulcerative colitis, it was shown that glycyrrhizin

significantly decreased the levels of TNF- $\alpha$ , IL-6, IL-8, and IL-17 and increased the expression of IL-10, which supports the anti-inflammatory effect of glycyrrhizin in the treatment of ulcerative colitis [70]. Glycyrrhizin also showed potential anti-inflammatory activity in inhibiting inflammation in patients with acne and cartilage inflammation in patients with osteoarthritis [71,72].

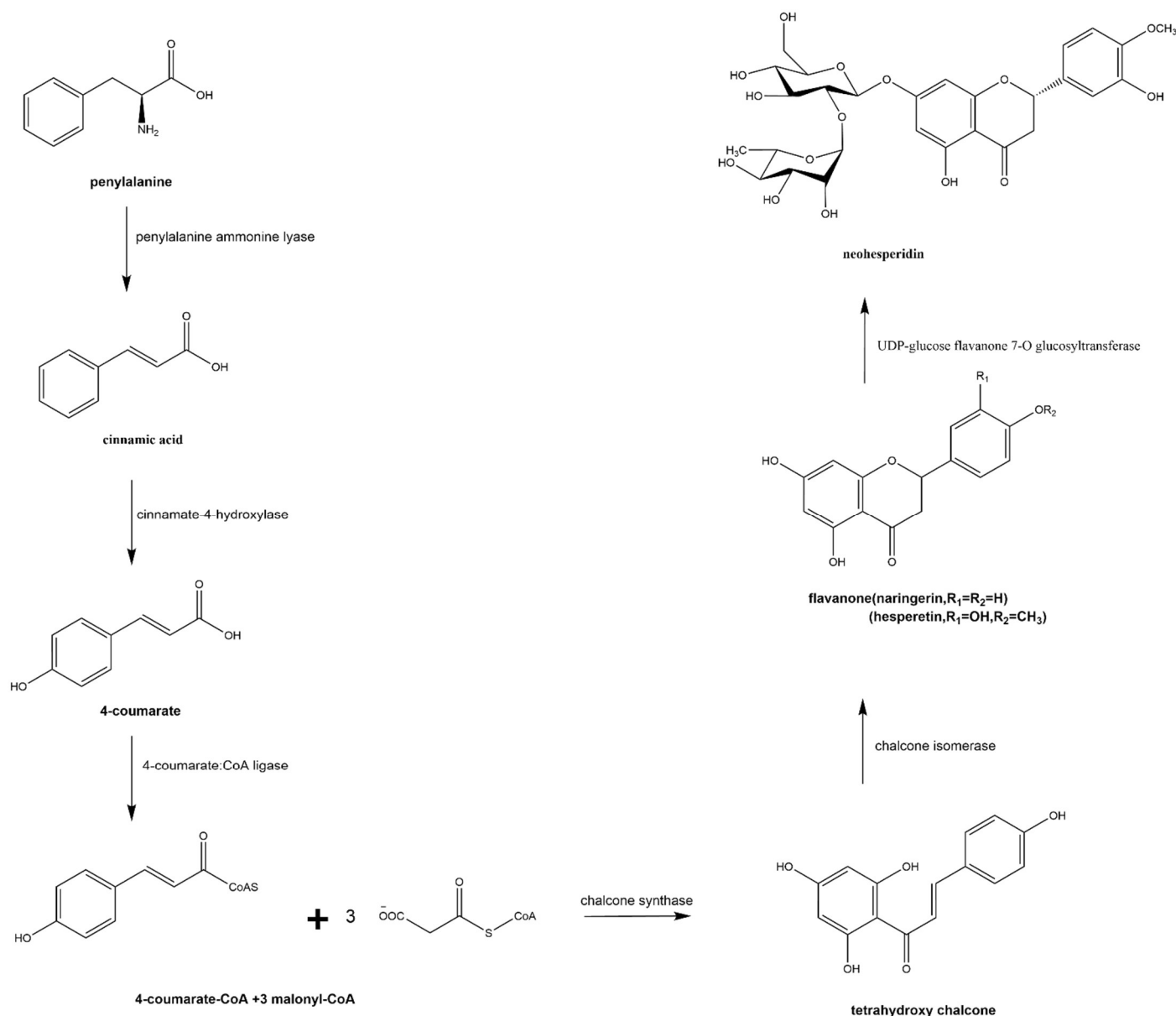
In conclusion, glycyrrhizin and glycyrrhetic acid demonstrate a wide range of pharmacological applications, including antioxidant, antiviral, anti-inflammatory, hepatoprotective, and antitumor activities. Their ability to enhance drug solubility and efficacy, coupled with their low toxicity, positions them as promising candidates for future drug development across multiple therapeutic areas.

## **5. Neohesperidin**

Neohesperidin, a semi-natural molecule from bitter orange. Treatment of neohesperidin with hot alkali leads to the opening of heterocyclic rings and the formation of the chalcone form of the flavanone glycoside [73]. Hydrogenation then takes place at the ethylenic double bond of the chalcone to produce the corresponding dihydrochalcones, neohesperidin dihydrochalcone (NHDC), which have an extremely high sweetening potential. Compared to aspartame, NHDC is stable even at high temperatures or under acidic/alkaline conditions, allowing it to remain food safe for up to five years when stored under optimal conditions [73]. Additionally, NHDC shows a synergistic sweetening effect when it is blended with a series of sweeteners [73]. In 2022, the European Food Safety Authority Panel on Food Additives and Flavorings conducted a re-evaluation of NHDC as a food additive, which proposed an acceptable daily intake of 20 mg/kg BW and the bioavailability of NHDC is 21.8% [73,74]. The NHDC market will grow at a CAGR of 4.6% from 2023 to 2030 [2].

### **5.1 Biosynthesis**

Numerous biogenic studies have demonstrated that neohesperidin is a flavonoid-derived secondary metabolite, synthesised through the phenylpropanoid pathway [75,76]. This pathway begins with the deamination of phenylalanine by the enzyme phenylalanine ammonia-lyase (PAL) to produce trans-cinnamic acid. Cinnamic acid is then subsequently hydroxylated by cinnamate 4-hydroxylase to form 4-coumarate. The 4-coumarate then undergoes CoA esterification via 4-coumarate-CoA ligase to form 4-coumaroyl-CoA. One molecule of p-coumaroyl-CoA then combines with three molecules of malonyl-CoA to yield naringenin chalcone, catalyzed by chalcone synthase [77,78]. Chalcone isomerase (CHI) facilitates the conversion of naringenin chalcone to naringenin, and later to hesperetin (a flavanone aglycone). Glycosylation adds a disaccharide to the C7 hydroxyl group of the aglycone, improving bioavailability, stability and solubility. Research shows that neohesperidin is synthesized via glycosylation of flavonoid aglycones such as naringenin, hesperidin and hesperetin, converting them into glucosides and rhamnoglucosides. This process is catalyzed by the enzymes UDP-glucose-flavanone 7-O-glucosyltransferase and UDP-rhamnose flavanone [79]. Neohesperidin is usually extracted from citrus fruits and purified from neohesperidose, a disaccharide responsible for its bitterness. It has also been observed that neohesperidin is converted to neohesperidin dihydrochalcone by catalytic hydrogenation under alkaline conditions.



**Figure 4.** The biosynthesis process of neohesperidin.

## 5.2 Bioactive

Neohesperidin has demonstrated significant biological effects in various research areas, particularly in anti-inflammatory, antioxidant and metabolic regulation. In lipid metabolism, neohesperidin regulates the PI3K/AKT/mTOR pathway, effectively inhibiting adipocyte proliferation and reducing lipid accumulation. Adipocyte formation was reduced by approximately 25–30%, demonstrating its potential for obesity control [80]. In addition, in metabolic disorders induced by a high-fat diet, neohesperidin significantly reduced serum triglyceride and cholesterol levels, with triglycerides reduced by 30% and cholesterol by 25% [81]. Neohesperidin dihydrochalcone also significantly reduced blood glucose levels in rats fed a high-fat diet by 10%, reduced serum triglycerides by 25% and increased the abundance of the beneficial bacterium *Coprococcus* by 40% [82]. In addition, in animal studies,

neohesperidine improved cholesterol metabolism by reducing total cholesterol by 25%, low-density lipoprotein (LDL) by 20% and high-density lipoprotein (HDL) by 15% [83].

In obesity-related studies, neohesperidin dihydrochalcone was found to reduce fat cell production by 30%, leading to a 20% reduction in body fat and a 10% increase in lean body mass, while significantly inhibiting the PI3K/AKT/mTOR pathway [84]. Moreover, the combination of neohesperidin and rebaudioside A reduced visceral fat by 40%, decreased liver lipid levels by 20% and promoted an increase in beneficial gut bacteria in obese mice [85].

In food preservation, neohesperidin acts as a natural antioxidant, effectively delaying lipid oxidation. The oxidation process was reduced by 50%, extending the shelf life of foods and improving sensory qualities [86]. In milk preservation, neohesperidin extended the shelf life by approximately 30% due to its antioxidant and antimicrobial properties [87].

Neohesperidin's anti-inflammatory properties have been demonstrated in models of inflammation in the digestive system. In an experimental colitis model, neohesperidin dihydrochalcone reduced inflammation levels by 45% by inhibiting the expression of IL-6 and TNF- $\alpha$ , while oxidative stress markers decreased by 50% [82]. In dextran sulphate sodium-induced ulcerative colitis, neohesperidine reduced IL-1 $\beta$ , IL-6 and TNF- $\alpha$  levels by 30–40%, improved intestinal barrier function and reduced intestinal permeability by 20% [88]. These anti-inflammatory effects were further supported in models of gastrointestinal inflammation, where neohesperidin reduced overall intestinal inflammation scores by more than 30% [89].

In bone health, neohesperidine has been shown to promote osteogenic differentiation of bone marrow stromal cells (BMSCs). By regulating histone modifications of the lncRNA SNHG1, it increased alkaline phosphatase activity by 50% and mineralised nodule formation by 60% [90]. Neohesperidine also activated the Wnt/ $\beta$ -catenin pathway, increasing ALP activity by 45% and mineralised nodule formation by 35% [91]. It also promoted the proliferation of BMSCs, with proliferation rates increasing by 30% and osteogenic differentiation markers improving by 45% [92].

Neohesperidin also exhibited potential in neurodegenerative disease research, as it was shown to inhibit amyloid protein aggregation, with an inhibition rate of 50%, while reducing oxidative stress and neuronal apoptosis, suggesting its application in conditions such as Alzheimer's disease [93]. Furthermore, in neuropathic pain models, neohesperidin alleviated mechanical hyperalgesia by regulating the P2X4 receptor, reducing pain scores by 50% [94].

Finally, in models of myocardial ischaemia-reperfusion injury, neohesperidine reduced cell apoptosis by 30% and oxidative stress (MDA) levels by 35% through inhibition of the JNK and NF- $\kappa$ B pathways [95].

In summary, neohesperidine exhibits a wide range of beneficial biological effects, particularly in lipid metabolism, anti-inflammation and bone health. Its ability to modulate key pathways, such as the PI3K/AKT/mTOR and Wnt/ $\beta$ -catenin pathways, together with its potent antioxidant and anti-inflammatory properties, positions it as a promising compound for the treatment of obesity, metabolic disorders, neurodegenerative diseases and inflammatory conditions. In addition, its potential

applications in food preservation and bone regeneration highlight its versatility and value across multiple therapeutic and industrial areas. Further research may unlock even more potential uses for neohesperidine, particularly in clinical settings.

## **6. Fructooligosaccharides**

Fructooligosaccharides (FOS), sometimes called oligofructose or fructooligosaccharides, are oligosaccharides used as an alternative sweetener. The sweetness of FOS is between 30% and 50% of the sugar in commercial syrups [96]. FOS can be extracted from the blue agave plant and from fruits and vegetables such as bananas, onions, chicory root, garlic, asparagus, jícama, and leeks. Some grains and cereals, such as wheat and barley, also contain FOS [97]. Given the resistance of FOS to digestion due to the absence of brush border  $\beta$ -fructosidases, it can act as an excellent bulking agent due to its low caloric content (1–1.5 kcal/g), which has the potential to reduce energy intake by 65-75% compared to digestible carbohydrates [98,99]. This also means that FOS reaches the colon intact and acts as a prebiotic for beneficial microorganisms in the gut (bifidobacteria being the most commonly targeted group), providing health benefits to the host [99].

### **6.1 Biosynthesis**

FOS synthesis is a complex biochemical process catalyzed by fructosyltransferases (FTases), which results in the formation of oligosaccharides with various structures. These oligosaccharides have wide applications in the food industry and biomedical fields due to their beneficial properties. FOS are short-chain oligosaccharides composed of sucrose and fructose units linked by different glycosidic bonds. Two key fructosyltransferases are involved in the synthesis of FOS: 1-FFT (1-fructosyltransferase) and 6-SFT (6-fructosyltransferase), each responsible for producing different types of FOS through different catalytic pathways.

1-FFT functions by transferring fructosyl groups from sucrose to another fructose or fructan polymer, forming new  $\beta(2\rightarrow1)$  glycosidic bonds. Sucrose serves as the initial substrate, and 1-FFT catalyzes the transfer of fructose from sucrose to another fructose molecule, producing the simplest form of FOS, 1-kestose. The molecular structure of 1-kestose consists of a glucose molecule and two fructose units linked by  $\beta(2\rightarrow1)$  glycosidic bonds. This step initiates the FOS synthesis process. As the reaction progresses, 1-FFT continues to add fructose units to 1-kestose, generating more complex FOS structures, such as 1-nystose, which contains additional fructose units. 1-nystose is often considered a precursor to inulin. Thus, 1-FFT plays a crucial role in the early stages of FOS synthesis, generating linear oligosaccharides linked by  $\beta(2\rightarrow1)$  bonds.

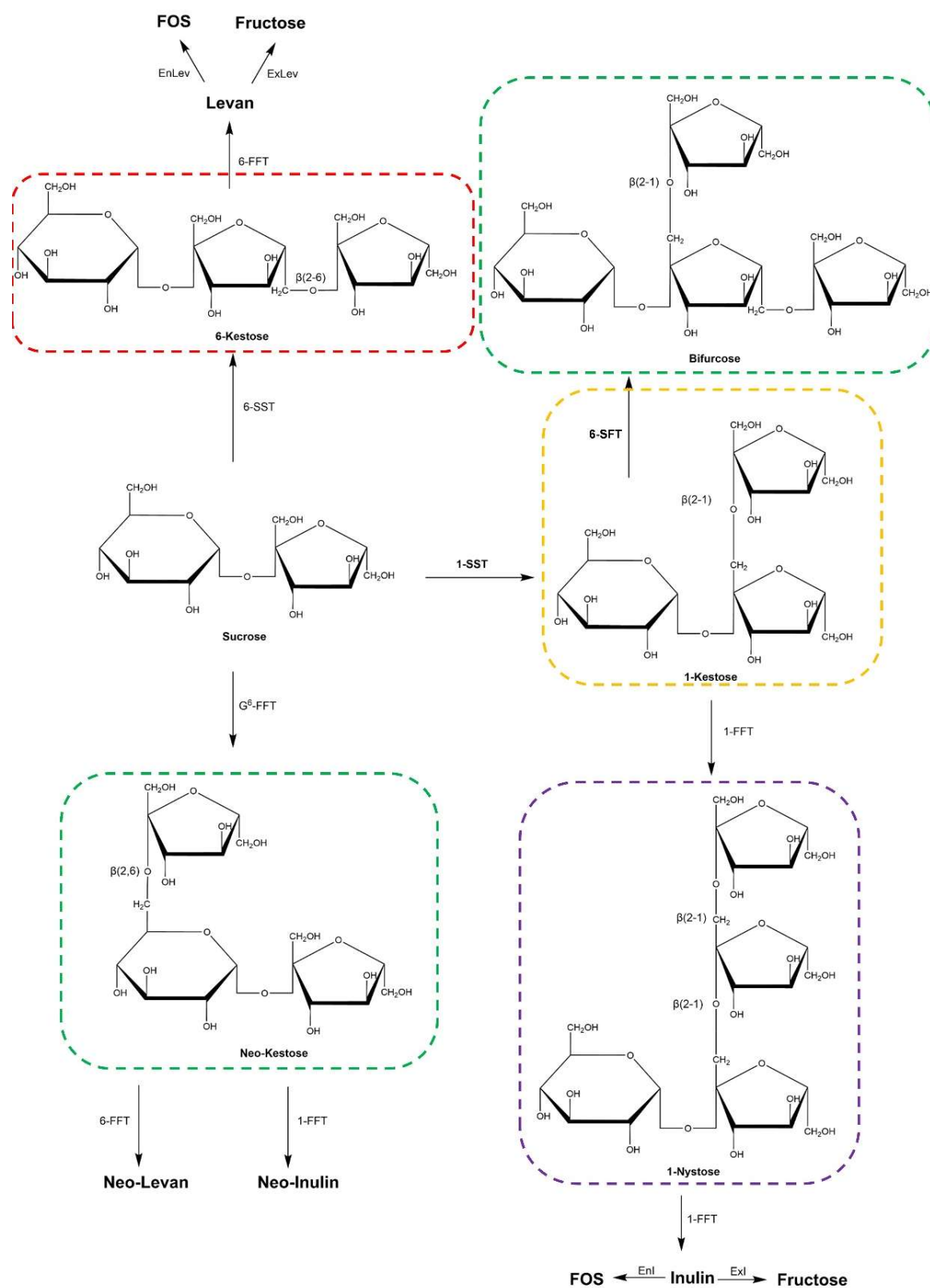
6-SFT catalyzes more complex reactions, involving the formation of both  $\beta(2\rightarrow1)$  and  $\beta(2\rightarrow6)$  glycosidic bonds, which contribute to more intricate FOS structures. Unlike 1-FFT, which primarily forms linear fructan chains, 6-SFT is capable of synthesizing branched FOS. By catalyzing both  $\beta(2\rightarrow1)$  and  $\beta(2\rightarrow6)$  linkages, 6-SFT creates more diverse and complex oligosaccharides. For example, bifurcose is a representative product of 6-SFT catalysis. Its structure contains both  $\beta(2\rightarrow1)$  and  $\beta(2\rightarrow6)$  bonds, formed by first attaching a fructose molecule to sucrose via a  $\beta(2\rightarrow1)$  bond, followed by the addition of another fructose molecule through a  $\beta(2\rightarrow6)$  bond,

resulting in a more branched structure. This dual function of 6-SFT allows for the generation of more structurally diverse FOS.

Throughout the FOS synthesis process, 1-FFT and 6-SFT work together to produce a variety of FOS types with different degrees of complexity. 1-FFT primarily generates linear fructooligosaccharides, such as 1-kestose and 1-nystose, which are linked by  $\beta(2\rightarrow1)$  bonds and form straight chains. In contrast, 6-SFT synthesizes both linear and branched FOS, such as bifurcose, by forming both  $\beta(2\rightarrow1)$  and  $\beta(2\rightarrow6)$  glycosidic bonds. The ability of 6-SFT to produce branched structures enhances the structural diversity of FOS and expands their potential applications.

Due to their unique structural properties, FOS synthesized by 1-FFT and 6-SFT have various functional applications. In the food industry, FOS are widely used as prebiotics that promote gut health by stimulating the growth of beneficial gut bacteria. Additionally, FOS serves as low-calorie sweeteners and dietary fiber supplements, providing health benefits without significantly impacting caloric intake. The ability to synthesize FOS with different structures also allows for the production of tailored oligosaccharides with specific functional properties suited for diverse industrial applications.

In conclusion, FOS synthesis is a process catalyzed by fructosyltransferases, which transfer fructose units to form oligosaccharides with various glycosidic bonds. 1-FFT mainly catalyzes the formation of FOS with  $\beta(2\rightarrow1)$  bonds, resulting in linear structures, while 6-SFT can form both  $\beta(2\rightarrow1)$  and  $\beta(2\rightarrow6)$  bonds, generating branched FOS with more complex structures. The combined action of these two enzymes produces a wide range of FOS with different structural and biological functions.



**Fig 5.** The biosynthesis process of fructooligosaccharides. 1-SST sucrose: sucrose 1-fructosyl-transferase; 1-FFT fructan: fructan 1-fructosyltransferase; 6-SST sucrose: sucrose 6-fructosyltransferase; 6-FFT fructan: fructan 6-fructosyltransferase, G6-FFT fructan: fructan 6G-fructosyltransferase; 6-SFT sucrose: fructan 6-fructosyltransferase; ExI exoinulinases; EnI endoinulinase; EnLev endolevanase; ExLev exolevanase



## 6.2 Bioactive

Fructooligosaccharides (FOS) have been extensively studied for their beneficial effects, particularly in relation to metabolic health, gut microbiota modulation, and inflammation reduction. FOS has been shown to promote the growth of beneficial gut bacteria, such as *Bifidobacterium* and *Lactobacillus*, while reducing harmful species like *Clostridium difficile*, resulting in improved gut health and reduced gastrointestinal discomfort [100,101]. This modulation of the gut microbiota also increases the production of short-chain fatty acids (SCFAs), particularly butyrate, which plays a key role in reducing inflammation and enhancing metabolic health [102,103].

In animal models, long-term FOS consumption has led to significant reductions in body weight and fat accumulation without changes in food intake, suggesting that FOS induces metabolic changes in the gut rather than limiting caloric intake [104,105]. Additionally, FOS has been found to reduce insulin resistance and systemic inflammation, demonstrating protective effects against metabolic syndrome. For example, in pregnant mice, FOS supplementation improved glucose tolerance and reduced obesity risk in their offspring, suggesting long-term metabolic benefits [106,107].

FOS also plays a crucial role in protecting liver health. Studies have shown that FOS supplementation reduces liver fat accumulation, improves lipid metabolism, and modulates gut-brain communication, offering protection against non-alcoholic fatty liver disease (NAFLD) [108,109]. Similarly, FOS has demonstrated protective effects on renal health, with studies showing reductions in oxidative stress and inflammatory markers in models of kidney injury, highlighting its potential as a therapeutic agent for kidney-related metabolic disorders [110].

Moreover, FOS shows synergistic effects when combined with other compounds. For example, the combination of FOS with galactooligosaccharides (GOS) enhances gut-brain axis signaling, while the combination of FOS and vitamin D3 significantly reduces intestinal inflammation and improves gut barrier function, suggesting broader therapeutic potential [111].

Studies have shown that ITF has a strong bifidobacterial effect on the human fecal microbiota and promotes the growth and activity of other bacteria known to have health benefits, such as *Lactobacillus* and *F. prausnitzii*. These effects on the gut microbiota contribute to human health and include improved intestinal barrier function, improved bowel movements, increased insulin sensitivity, lower blood triglycerides and improved lipid profiles, increased calcium and magnesium absorption, and increased satiety [112].

Collectively, these findings highlight the broad applications of FOS in managing metabolic diseases, promoting gut health, reducing inflammation, and potentially treating liver and kidney disorders. FOS is emerging as a valuable dietary supplement with both preventive and therapeutic applications, contributing to improved health outcomes in diverse populations [113,114].

## 7. Conclusion

In conclusion, natural sweeteners such as steviol glycosides, mogrosides, glycyrrhizin, neohesperidin, and fructooligosaccharides offer significant promise as

healthier alternatives to traditional sugars and artificial sweeteners. Their low or zero-calorie content, combined with various bioactive effects—including antioxidant, anti-inflammatory, and anti-diabetic properties—make them valuable in managing metabolic disorders like obesity and diabetes. The study of their biosynthetic pathways and pharmacological activities not only underscores their potential to promote human health but also paves the way for broader applications in the food, pharmaceutical, and nutraceutical industries. With the rising demand for functional, health-promoting ingredients, these natural sweeteners are well-positioned to play a key role in the development of health-conscious products. Future research should focus on optimizing their production methods and further exploring their long-term health effects to fully capitalize on their therapeutic potential.

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